Supporting Information

for

Unexpected chiral vicinal tetrasubstituted diamines via borylcopper-mediated homocoupling of isatin imines

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General Information

$^1$H NMR and $^{13}$C NMR spectra were recorded using a Bruker AV 400 Ultrashield spectrometer. $^1$H NMR and $^{13}$C NMR chemical shifts were reported in parts per million (ppm) downfield from tetramethylsilane. Coupling constants ($J$) were reported in Hertz (Hz). The residual solvent peaks were used as internal reference: $^1$H NMR (CD$_3$CN 1.98 ppm) $^{13}$C NMR (CD$_3$CN 0.3 ppm, 117.3 ppm). The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, m= multiplet, br= broad. All the N-Substituted isatins$^1$ and the corresponding ketimines were synthetized according to the previous literature and their spectroscopic data are in agreement with the reported ones.$^2,3$ Other reagents were received from commercial sources (Fluorochem, TCI and Merk) and used without further purifications. Column chromatography were performed by Flash Chromatography (FC) using Merk Silica gel 60. When the reaction was scaled up, traces of pinacol persists after FC and can be seen in NMR spectra ($^1$H NMR 400 MHz, CD$_3$CN δ 4.16 br s, 1.15 s; $^{13}$C NMR 101 MHz, CD$_3$CN δ 22.1, quaternary carbons not observed).

General Procedure for the synthesis of compounds 2

In a round-bottom flask, (Ph)$_3$P (0.025 eq.) was dissolved in dry toluene (0.024 M). Saturated aqueous solution of copper sulfate pentahydrate (0.025 eq, 0.78M) was added followed by DIPEA (0.5 eq.) and the biphasic mixture stirred at high speed (1500 rpm) for 10 minutes (aqueous solution goes from a dark-blue to a pale blue colour). B$_2$pin$_2$ (1.5 eq.) was added followed by a solution of ketimine 1 in dry toluene (1.0 eq, 0.47M) and the reaction stirred at high speed and room temperature until the disappearance of 1 (monitored by TLC, approx. 5h in which the reaction goes from red/orange to brown). The reaction was diluted with dichloromethane and filtered over a short pad of celite, then the solvents were removed under reduced pressure. The mixture was diluted with dichloromethane, washed with brine, dried over anhydrous Na$_2$SO$_4$ and the solvent removed under reduced pressure. The crude mixture was purified by FC to afford pure compound 2.

(R,R)-N,N’-((3R,3’S)-1,1'-dimethyl-2,2'-dioxo-[3,3’-biindoline]-3,3'-diyl)bis(2-methylpropane-2-sulfinamide) (2a)

Synthetized according to the General Procedure starting from N-Me isatin-derived ketimine 1a; purified by FC (Dichloromethane / Ethyl acetate 8:2 to 6:4) to afford a salmon pink solid (yield 68%); [$\alpha$]$_D$ = -136.1 (C= 1.0 in CHCl$_3$), melting point: 152-156°C, $^1$H NMR (400 MHz, CD$_3$CN) δ 8.11 (d, $J$= 7.6 Hz, 1H), 7.58 (t, $J$= 7.6 Hz, 1H), 7.35-7.25 (m, 2H), 6.96-6.89 (m, 2H), 6.88 (s, 1H), 6.68 (t, $J$= 7.6 Hz, 1H), 6.16 (s, 1H), 5.69 (d, $J$= 7.6 Hz, 1H), 3.18 (s, 3H), 2.69 (s, 3H), 1.26 (s, 9H), 1.21 (s, 9H); $^{13}$C NMR (101 MHz, CD$_3$CN) δ 176.7 (1C), 172.9 (1C), 145.8 (2C), 132.1 (1C), 131.6 (1C), 130.3 (1C), 126.2 (1C), 122.67 (1C), 122.57 (1C), 121.8 (2C), 109.7 (1C), 109.5 (1C), 70.0 (1C), 63.8 (1C), 57.1 (1C), 56.7 (1C), 26.7 (1C), 26.0 (1C), 22.79 (3C), 22.76 (3C); HRMS-ESI [M+Na]$^+$ calculated for C$_{26}$H$_{34}$N$_4$O$_4$S$_2$Na$^+$ 553.1919 found 553.1924.
(R,R)-N,N’-((3R,3’S)-1′,1′-dibenzyl-2,2′-dioxo-[3,3′-biindoline]-3,3′-diyl)bis(2-methylpropane-2-sulfinamide) (2b)

Synthesized according to the General Procedure starting from N-Bn isatin-derived ketimine 1b; purified by FC (Dichloromethane / Ethyl acetate 8:2 to 7:3) to afford a purple foam (yield 72%); [α]D = −126.1 (C=1.0 in CHCl3); 1H NMR (400 MHz, CD3CN) δ 8.18 (d, J= 7.4 Hz, 1H), 7.63-7.55 (m, 2H), 7.45 (t, J= 7.4 Hz 1H), 7.36-7.22 (m, 5H), 7.20-7.07 (m, 3H), 6.99 (s, 1H), 6.77 (d, J= 7.6 Hz, 1H), 6.69 (d, J= 7.6 Hz, 1H), 6.65-6.55 (m, 3H), 6.30 (s, 1H), 5.84 (d, J= 7.6 Hz, 1H), 5.11 (d, J= 16.1 Hz, 1H), 4.81 (d, J= 16.1 Hz, 1H), 4.79 (d, J= 16.1 Hz, 1H), 4.42 (d, J= 16.1 Hz, 1H), 1.31 (s, 9H), 1.25 (s, 9H); 13C NMR (101 MHz, CD3CN) δ 177.3 (1C), 173.3 (1C), 145.2 (1C), 144.7 (1C), 136.5 (1C), 135.9 (1C), 131.9 (1C), 131.7 (1C), 130.3 (1C), 129.4 (2C), 129.2 (2C), 128.1 (3C), 128.0 (1C), 127.2 (2C), 126.9 (1C), 123.1 (1C), 122.8 (1C), 122.1 (2C), 110.9 (1C), 110.6 (1C), 69.8 (1C), 63.7 (1C), 57.3 (1C), 56.9 (1C), 44.7 (1C), 43.8 (1C), 22.9 (3C), 22.7 (3C); HRMS-ESI [M+Na]+ calculated for C38H42N4O4S2Na+ 705.2545 found 705.2540.
Copies of $^1$H and $^{13}$C NMR spectra

$^1$H NMR spectrum of compound 2a (400 MHz, CD$_3$CN)

$^{13}$C NMR spectrum of compound 2a (100 MHz, CD$_3$CN)
$^1$H NMR spectrum of compound 2b (400MHz, CD$_3$CN)

$^{13}$C NMR spectrum of compound 2b (100 MHz, CD$_3$CN)
Crystallographic data for compound 2a

The sample selected for the X-ray analysis was a yellowish, transparent prism with dimensions ≈ 0.500 x 0.350 x 0.175 mm (Figure S1). It was mounted on a glass fibre, using perfluorinated oil as a glue.

The data collection was carried out with a Bruker AXS Smart Apex three-circle diffractometer, equipped with a CCD area detector. Graphite-monochromated Mo Kα radiation (λ=0.71073 Å) was employed at a nominal power of 50 kV x 30 mA of the X-ray source. The data collection consisted in 5 redundant ω-scans at steps of 0.25 deg for an exposure time of either 10 s/frame or 20 s/frame, all performed in ambient conditions. Eventually, we explored a 98.8 % complete full sphere of data, up to a maximum sin(θ/λ) of 0.72 Å⁻¹. The raw dataset consisted of 38838 measured reflections, 91 of which were rejected as systematic absence violations. A negligible twin component was present, which was ignored in the subsequent data reduction. The data were integrated with the commercial SAINT+ program and corrected for absorption (μ = 0.230 mm⁻¹) and anisotropic beam intensity with SADABS. The final dataset consisted of 8556 independent reflections (5979 with I > 2σ(I)) with R_int = 0.0373. The space group was unequivocally established by systematic absence relationships.

The substance 2a crystallizes in P2₁2₁2₁ (system orthorhombic), with unit cell parameters a = 10.826(2) Å, b = 13.862(2) Å, c = 18.387(3) Å, V = 2759(1) Å³, density 1.277 g/cm³. The structure was solved by direct methods by shelxs and refined by least squares within the independent atom model approximation implemented in shelxl. The final crystallographic agreement factors were Goodness-of-fit: 1.085, R1(F) = 0.0547 for 5979 F₀ > 4σ(F₀), and R1(F) = 0.0802, wR2(F²)= 0.1634 for all the 8556 independent data. Maximum and minimum Fourier residuals read as ΔρMAX/MIN = +0.20/ –0.55 e/Å³.
**Figure S2.** Wires–stick representation of the crystal packing of 2a at RT, as seen along the a (a), b (b) and c (c) cell axes. Colour code as in Figure 1 (main text).

**Discussion**

The compound is chiral and crystallizes in the orthorhombic Sohncke space group $P2_12_12_1$ with 4 formulae in cell and 1 molecule in the asymmetric unit. The absolute configuration of the two S stereogenic centres is confirmed to be R (see main text). The Flack parameter, computed by classical fits against all the intensities$^7,8$, refines to 0.1(1).

The C2–C10 single bond (numbering scheme in Figure 1, main text) is quite long (1.5863(2) Å), as expected due to crowding of the two facing oxindole systems. In the crystal, the mean least-squares planes across the C-N backbones of the two indole rings are mutually rotated by ~48.6 deg. At the same time, NH groups set up intramolecular hydrogen bonds with the O acceptors of the sulfinamide moieties (see main text). Likely, such hydrogen bonds are crucial to stabilize the conformer despite a minor intramolecular H3N···H21C steric clash of the N3 amine with the C21 methyl ($d_{\text{H···H}} = 1.89$ Å).

Figure 2 shows the main packing motifs in the (b,c), (a,c) and (a,b) planes. The only relevant hydrogen bond donors are the NH moieties in the sulfinamide groups, while both S=O and the keto group in the 2-oxindole rings can serve as acceptors. However, both donors are saturated by intramolecular hydrogen bonded contacts (Table 1, main text), resulting in no extended HB patterns through the crystal. Also, no relevant stacking interactions can be found, due to the misalignment of the neighbouring oxindole systems (see above). Consequently, the main structure-determining interactions are expected to be van der Waals and electrostatic interactions.
References