1.	Reagents and instrumentation	S-2
2.	Synthetic procedures and characterization data for novel compounds	S-3
3.	Attempted oxidation of difluorinated alcohol 13	S-12
4.	Measurement of enantiopurity	S-13
5.	Reproductions of NMR spectra	S-14
6.	Photostability assessment	S-34
7.	Conformational analysis of compound 2	S-36
8.	Biological assays	S-40
9.	References	S-42

1. Reagents and instrumentation

Reactions were performed in oven-dried glassware at room temperature and under a nitrogen atmosphere with magnetic stirring unless stated otherwise. All commercial reagents were of synthetic grade and used as received. Dry solvents were obtained by passage through an alumina column, and/or by storage over 4Å molecular sieves. Reactions were monitored by thin layer chromatography (TLC) using Merck aluminum-backed silica gel 60 F₂₅₄ (0.2mm) TLC plates. TLC product spots were visualised either under short-wave UV light (254 nm) or by staining with KMnO₄ and heating. Flash chromatography was performed using Davisil 40-63 mesh silica gel, and eluents are stated as volume-to-volume ratios. Melting points were determined using an OptiMelt melting point apparatus MPA100. IR spectra were recorded using a Cary 630 Fourier Transform Infrared Spectrometer equipped with attenuated total reflectance (ATR) with a diamond crystal inset. NMR spectra were obtained using Bruker Advance III 300, 400 and 600 MHz instruments at 300 K. Residual solvent peaks were used as an internal reference to calibrate ¹H and ¹³C spectra. Splitting patterns are represent by s = singlet, d = doublet, br s = broad singlet, dm = doublet of multiplets (typically for CHF protons where ${}^{2}J_{\text{HF}}$ is easy to determine while the others are difficult). When necessary to aid in the peak assignment, 2D experiments including COSY, HMBC and HSQC were acquired. Coupling constants from complex spectra were obtained by Daisy simulation using the Bruker TopSpin software. HRMS results were acquired at the UNSW Bioanalytical Mass Spectrometry Facility using an Orbitrap LCQ XP Plus ion trap MS operating in positive ion mode using electrospray ionization (ESI). Optical rotations were measured using a Perkin Elmer model 341 polarimeter (λ = 589 nm; l = 1 dm and c expressed in grams per 100 mL).

2. Synthetic procedures and characterization data for novel compounds

(2R,3R)-5-(Benzo[d][1,3]dioxol-5-yl)-2,3-difluoro-1-(piperidin-1-yl)pentan-1-one (2)

solution of compound 20 (12.5 mg, 0.048 mmol), 1-А hydroxybenzotriazole monohydrate (80%, 8.20 mg, 0.048 mmol) and N,N'-diisopropylcarbodiimide (0.048 mmol) in dry DMF (1.0 mL) was stirred at r.t for 15 min. Piperidine (4.31 mg, 0.508 mmol) was added, and the mixture was stirred at r.t. under nitrogen for 22 h. The mixture was evaporated under reduced pressure and the residue was dissolved in ethyl acetate (30 mL). The organic layer was washed with citric acid (15 mL), sodium bicarbonate (15 mL) and brine (15 mL), then dried (Na₂SO₄) and concentrated onto silica. The crude product was subjected to flash chromatography eluting with $5:95 \rightarrow 40:60$ ethyl acetate/hexane to afford the title compound as a yellow oil (8.1 mg, 51%); $[\alpha]_{\mathbf{D}}$ +35.7 (c 0.385, CHCl₃); IR $(DCM) v_{max} (cm^{-1}) 3257, 2940, 1643, 1490, 1445, 1368, 1246, 1189, 1101, 1038; {}^{1}H NMR$ $(600 \text{ MHz}, \text{CDCl}_3) \delta 6.73 \text{ (d}, J = 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 6.68 \text{ (d}, J = 1.6 \text{ Hz}, 1\text{H}, \text{ArH}), 6.65 \text{ (dd}, J = 1.6, 1000 \text{ Hz})$ 7.9 Hz, 1H, ArH), 5.92 (s, 1H, OCH₂O), 5.04 (ddd, J=3.5, 25.2, 49.0, 1H, CHFCHFCO), 4.83 (dddt, J = 3.5, 9.8, 22.1, 47.8, 1H, CHFCHFCO), 3.64–3.47 (m, 4H, CH₂NCH₂), 2.79 (ddd, J = 4.9, 8.9, 14.3, 27.6 Hz, 1H, CH₂CHHCHF), 1.90 (ddddd, *J* = 3.5, 8.2, 9.2, 14.3, 35.1 Hz, 1H, CH₂CHHCHF), 1.67–1.53 (m, 6H, N–CH₂CH₂CH₂), ${}^{13}C{}^{1}H{}$ NMR (151 MHz, CDCl₃) δ 164.6 (dd, 5.3, 18.7 Hz), 147.9, 146.1, 134.3, 121.5, 109.1, 108.5, 101.0, 92.0 (dd, J = 20.1, 196.8 Hz), 91.2 (dd, J = 19.4, 177.6 Hz), 46.6 (d, J = 12.2 Hz), 44.2, 32.5 (dd, J = 4.2, 21.1 Hz), 30.8 (d, J = 4.1 Hz) 26.7, 25.8, 24.6; ¹⁹F NMR (377 MHz, CDCl₃), δ –194.2 (s, 1F, CH₂CH₂CHF), –195.8 (s, 1F, CH₂CHFCHF); ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ –194.2 (d, J = 12.3 Hz, CH₂CH₂CH<u>F</u>), –195.8 (d, J = 12.3 Hz, CH₂CHFCH<u>F</u>); HRMS (ESI, +ve) $C_{17}H_{21}O_3F_2N_1Na^+$ [MNa⁺] requires m/z 348.1382, found 348.1372.

(2*S*,3*S*)-5-(benzo[*d*][1,3]dioxol-5-yl)-1-(benzyloxy)pentane-2,3-diol (8)

Step 1: To a suspension of sodium hydride (60% dispersion in mineral oil, 48.2 mg, 1.26 mmol) in dried THF (2 mL) at 0 °C was added allylic alcohol $7^{[1]}$ (129 mg, 0.627 mmol), and the reaction mixture was stirred at 0 °C under

nitrogen for 1 hour. Benzyl bromide (323 mg, 1.89 mmol) was added, and the mixture was stirred at

r.t. for 24 h. Mixture was quenched with saturated sodium bicarbonate and extracted with ethyl acetate (5 × 50 mL), and the combined organic layers were dried (MgSO₄) and concentrated onto silica. The residue was subjected to column chromatography, eluting with 1:4 ethyl acetate/hexane, to afford the corresponding benzyl ether as a clear oil (106 mg, 57%); **IR** (DCM) v_{max} (cm⁻¹) 2922, 2851, 2359, 1732, 1606, 1487, 1440, 1360, 1243, 1186, 1096, 1039; ¹H NMR (400 MHz, CDCl₃) δ 7.38–7.28 (m, 5H, ArH), 6.72 (d, *J* = 7.9 Hz, 1H. ArH), 6.68 (d, *J* = 1.5 Hz, 1H. ArH), 6.63 (dd, *J* = 1.5, 7.9 Hz, 1H. ArH), 5.91 (s, 2H, OCH₂O), 5.73 (dt, *J* = 6.3, 15.4 Hz, 1H, CH=CHCH₂OBn), 5.62 (dt, *J* = 6.4, 15.4 Hz, 1H, CH=CHCH₂OBn), 4.48 (app s, 2H, PhCH₂), 3.97 (s, 1H, CHHOBn), 3.96 (s, 1H, CHHOBn), 2.63 (dd, *J* = 7.2, 8.2 Hz, 2H, CH₂CH₂), 2.34 (dd, *J* = 7.2, 7.7 Hz, 2H, CH₂CH₂); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 147.6, 145.7, 138.6, 135.7, 133.8, 128.5, 127.9, 127.7, 127.2, 121.2, 109.0, 108.2, 100.9, 72.0, 70.9, 35.4, 34.5; HRMS (ESI, +ve) C₁₉H₂₀O₃Na⁺ [MNa⁺] requires *m/z* 319.1305, found 319.1297.

Step 2: A mixture of sodium bicarbonate (89.3 mg, 1.06 mmol), potassium carbonate (147 mg, 1.06 mmol), potassium ferricyanide (350 mg, 1.06 mmol), (DHQ)₂PHAL (2.80 mg, 0.004 mmol), tert-butanol (880 µL) and water (880 µL) was stirred at r.t for 30 min then cooled to 0 °C. Osmium tetroxide (0.5% solution in water, 0.54 mL, 0.01 mmol) and methanesulfonamide (33.6 mg, 0.354 mmol) were added, and the reaction was stirred at 0 °C for 1h. The benzyl ether synthesised above (105 mg, 0.354 mmol) was added, and the mixture was stirred at 0 °C for 24 h. A second portion of osmium tetroxide (0.5% solution in water, 0.54 mL, 0.01 mmol) was added, and the mixture was stirred for an additional 24 h. Sodium sulfite (560 mg, 4.44 mmol) was added, and the reaction was stirred at r.t for 1 h. The mixture was extracted with ethyl acetate (4 × 20 mL) and dichloromethane (4 × 25 mL), and the combined organic layers were dried (MgSO₄) and concentrated onto silica. The crude product was subjected to flash chromatography eluting with 1:4 ethyl acetate/hexane to give the title compound as a clear waxy solid (78.1 mg, 67%); $[\alpha]_{\rm D}$ -26.4 (c 1.06, CHCl₃); **m.p.** 85.6–85.9 °C; **IR** (neat) v_{max} (cm⁻¹) 3286, 3028, 2895, 2863, 2318, 1607, 1360, 1484, 1438, 1360, 1239, 1106, 1036, 1004; ¹H NMR (400 MHz, CDCl₃) δ 7.37–7.29 (m, 5H, ArH), 6.72 (d, J = 7.7 Hz, 1H, ArH), 6.69 (d, J = 1.2 Hz, 1H, ArH), 6.64 (dd, J = 1.2, 7.7 Hz, 1H, ArH), 5.91 (s, 1)2H, OCH₂O), 4.56 (d, J = 11.8 Hz, 1H, CHHPh), 4.52 (d, J = 11.8 Hz, 1H, CHHPh), 3.63–3.54 (m, 4H, CHOH-CHOH-CH₂OBn), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, CHHCH₂CHOH), 2.74 (ddd, *J* = 5.4, 9.1, 14.5, 1H, 14.5, 1H 9.1, 14.5, 1H, CHHCH₂CHOH), 2.61 (ddd, J = 7.5, 9.1, 13.6, 1H, CHHCH₂CHOH), 1.83 (ddd, J =5.3, 13.4, 14.1, 1H, CH₂C<u>H</u>HCHOH), 1.70 (ddd, J = 3.4, 6.9, 13.4, 1H, CH₂C<u>H</u>HCHOH); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 147.7, 145.7, 137.7, 135.9, 128,7, 128.1, 127.9, 121.3, 109.1, 108.3, 100.9, 72.8, 72.6, 71.4, 35.5, 31.6; **HRMS** (ESI, +ve) $C_{19}H_{22}O_5Na^+$ [MNa⁺] requires m/z353.1359, found 353.1359.

(4*S*,5*S*)-4-(2-(Benzo[*d*][1,3]dioxol-5-yl)ethyl)-5-((benzyloxy)methyl)-1,3,2-dioxathiolane 2,2dioxide (9)



Step 1: Thionyl chloride (0.033 mL, 0.451 mmol) was added to a solution of the diol **8** (74.4 mg, 0.225 mmol) and pyridine (0.055 mL, 0.676 mmol) in dichloromethane (2.6 mL) at 0 °C, and the resulting mixture was stirred at 0 °C for 1 h. Saturated aqueous copper sulfate (20.0 mL) was added, and the reaction

was extracted with dichloromethane ($3 \times 40 \text{ mL}$). The combined organic layers were dried (MgSO₄) and concentrated to give the cyclic sulfite as yellow/green oil, which was used without purification for the next step.

Step 2: The crude cyclic sulfite (~0.225 mmol) was dissolved in acetonitrile (1.78 mL) and dichloromethane (1.78 mL) and the solution was cooled to 0 °C. Sodium metaperiodate (96.4 mg, 0.451 mmol), ruthenium chloride hydrate (~10 mg) and water (2.62 mL) were added, and the mixture was stirred at 0 °C for 2 h. Diethyl ether (40 mL) was added, and the mixture was washed with water $(2 \times 75 \text{ mL})$, saturated aqueous sodium bicarbonate $(2 \times 75 \text{ mL})$ and brine $(2 \times 75 \text{ mL})$. The organic layer was dried (MgSO₄) and concentrated to give the title compound as a dark orange oil (79.5 mg, 90%). $[\alpha]_{\rm D}$ -66.7 (c 0.150, CHCl₃); **IR** (DCM) $v_{\rm max}$ (cm⁻¹): 2920, 1742, 1489, 1444, 1384, 1208, 1100, 1038; ¹H NMR (400 MHz, CDCl₃) δ 7.34–7.25 (m, 5H, ArH), 6.74 (d, J = 7.9 Hz, 1H, ArH), 6.65 $(d, J = 1.4 \text{ Hz}, 1\text{H}, \text{ArH}), 6.61 (dd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (ddd, J = 1.4, 7.9 \text{ Hz}, 1\text{H}, \text{ArH}), 5.94 (s, 2\text{H}, \text{OCH}_2\text{O}), 4.79 (s, 2\text{H}, \text{$ 3.1, 7.6, 9.8 Hz, 1H, CHSO₄CH₂CH₂OBn), 4.65 (dt, *J* = 4.7, 7.6 Hz, 1H, CHSO₄CHCH₂OBn), 4.59 $(d, J = 11.8 \text{ Hz}, 1\text{H}, C\underline{H}HPh), 4.54 (d, J = 11.8 \text{ Hz}, 1\text{H}, CH\underline{H}Ph), 3.74 (dd, J = 4.6, 11.0 \text{ Hz}, 1\text{H}, 10.0 \text{ Hz})$ CHHOBn), 3.70 (dd, J = 4.6, 11.0 Hz, 1H, CHHOBn), 2.80 (ddd, J = 4.9, 9.0, 13.9 Hz, 1H, CHHCH₂CHSO₄), 2.66 (dt, J = 8.3, 13.9 Hz, 1H, CHHCH₂CHSO₄), 2.18 (dddd, J = 4.9, 8.5, 9.4, 14.5 Hz, 1H, CHHCH₂CHSO₄), 1.99 (dddd, J = 3.2, 8.0, 8.5, 14.5 Hz, 1H, CHHCH₂CHSO₄); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 148.1, 146.4, 136.8, 133.1, 128.8, 128.4, 128.0, 127.9, 121.5, 108.9, 108.6, 101.1, 84.3, 83.7, 74.0, 67.3, 34.5, 31.1; HRMS (ESI, +ve) C₁₉H₂₁O₇S⁺ [MH⁺] requires *m*/*z* 393.1003, found 393.1002.

(2S,3R)-5-(Benzo[d][1,3]dioxol-5-yl)-1-(benzyloxy)-3-fluoropentan-2-ol (10)



A mixture of tetra-*N*-butylammonium fluoride (1M solution in THF, 10 mL, 10 mmol) and acetonitrile (60 mL) was dried over 4Å molecular sieves then added via cannula to an ice-cold flask containing cyclic sulfate **8** (1.86 g,

4.75 mmol), and the resulting mixture was stirred at r.t. for 18 h. The mixture was concentrated to

give a clear brown oil. This residue was dissolved in a mixture of sulfuric acid (810 µL), water (148 µL) and tetrahydrofuran (148 mL), and the resulting solution was stirred at r. t., for 17 h. Brine (150 mL) and water (150 mL) were added, and the mixture was extracted with dichloromethane (5 × 300 mL). The combined organic layers were dried (MgSO₄) and concentrated onto silica. The crude product was subjected to flash chromatography eluting with 3:7 ethyl acetate/hexane to give the title compound as a yellow liquid (551 mg, 35%); $[\alpha]_{\rm D}$ +16.5 (*c* 1.34, CHCl₃); **IR** (DCM) $v_{\rm max}$ (cm⁻¹) 3045, 2920, 2350, 1734, 1488, 1441, 1363, 1263, 1095, 1036; ¹H NMR (400 MHz, CDCl₃) δ 7.38– 7.29 (m, 5H, ArH), 6.73 (d, J = 7.9 Hz, 1H, ArH), 6.69 (d, J = 1.3 Hz, 1H, ArH), 6.65 (dd, J = 1.3, 7.9 Hz, 1H, ArH), 5.92 (s, 2H, OCH₂O), 4.58 (d, 1H, J = 12.3 Hz, 1H, CHHPh), 4.54 (d, 1H, J = 12.3 Hz, 1H, CHHPh), 4.46 (ddd, J = 6.4, 12.3, 48.5 Hz, 1H, CFH), 3.84 (ddd, J = 3.6, 6.5, 10.2 Hz, 1H, CHOH), 3.65 (ddd, J = 2.4, 3.3, 9.6 Hz, 1H, CHHOBn), 3.57 (ddd, J = 1.5, 6.4, 9.6 Hz, 1H, CHHOBn), 2.80 (ddd, J = 6.1, 8.3, 14.4 Hz, 1H, CHHCH₂CHF), 2.63 (dt, J = 8.2, 14.4 Hz, 1H, CHHCH₂CHF), 2.08–1.91 (m, 2H, CH₂CH₂CHF); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 147.7, 145.9, 137.8, 135.2, 128.7, 128.7, 127.9, 121.4, 109.1, 108.3, 100.9, 92.5 (d, *J* = 170.9 Hz), 73.7, 71.6 (d, *J* = 24.3 Hz), 70.4 (d, J = 5.2 Hz), 33.3 (d, J = 20.4 Hz), 30.1 (d, J = 3.2 Hz); ¹⁹F NMR (377 MHz, CDCl₃) δ -193.6 (s, 1F); ¹⁹F{¹H} NMR (377 MHz, CDCl₃) δ -193.6; HRMS (ESI, +ve) $C_{19}H_{21}FO_4Na^+$ [MNa⁺] requires *m/z* 355.1316, found 355.1317.

5-((3*R*,4*R*)-5-(benzyloxy)-3,4-difluoropentyl)benzo[*d*][1,3]dioxole (11)

A mixture of fluorohydrin 10 (99.4 mg, 0.299 mmol) and a solution of Deoxo-FluorTM in toluene (2.7 M, 1.1 mL, 3.0 mmol) was stirred at 70 °C in a Teflon vessel for 24 h. The mixture was cooled to 0 °C, diluted with dichloromethane, and concentrated onto silica. The crude product was subjected to flash chromatography eluting with 1:9 ethyl acetate/hexane to give the title compound as a pale yellow oil (25.4 mg, 25%) and the side product 12 (21.2 mg, 23%). Data for 11: [α]_D+15.3 (*c* 1.26, CHCl₃); IR (DCM) v_{max} (cm⁻¹) 3028, 2923, 2872, 1866, 1607, 1488, 1442, 1364, 1246, 1112, 1039; ¹H NMR (400 MHz, CDCl₃) δ 7.38–7.28 (m, 5H, ArH), 6.74 (d, J = 7.8 Hz, 1H, ArH), 6.68 (d, J = 1.2 Hz, 1H, ArH), 6.64 (dd, J = 1.2, 7.8 Hz, 1H, ArH), 5.93 (s, 1H, OCH₂O), 4.75–448 (m, 2H, CHFCHF), 4.57 (app s, 2H, PhCH₂O), 3.76 (d, J = 5.0 Hz, 1H, CHHOBn), 3.71 (dd, J = 2.5, 5.0 Hz, 1H, CHHOBn), 2.77 (ddd, J = 5.0, 9.0, 14.0 Hz, 1H, CHHCH₂CHF), 2.65 (dt, J = 8.1, 14.0 Hz, 1H, C<u>H</u>HCH₂CHF), 2.11 (dddd, J = 5.0, 9.0, 27.8 Hz, 1H, CH₂C<u>H</u>HCHF), 1.83 (ddddd, J = 3.4, 7.4, 14.0, 1.0034.8 Hz, 1H, CH₂CHHCHF); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 147.8, 146.0, 137.6, 134.6, 128.6 ,128.0, 127.9, 121.4, 109.0, 108.4, 101.0, 92.3 (dd, J = 19.7, 179.0 Hz), 90.7 (dd, J = 19.8, 175.6 Hz),

73.8, 68.5 (dd, J = 6.9, 24.8 Hz), 32.3 (dd, J = 4.9, 21.1 Hz), 30.8 (d, J = 4.4 Hz); ¹⁹F NMR (377 MHz, CDCl₃), δ –200.3 (s, 1F, CH₂CH₂CH<u>F</u>), –202.1 (s, 1F, CH₂CH₂CHFCH<u>F</u>); ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ –200.3 (d, J = 10.5 Hz, CH₂CH₂CHF), –202.1 (d, J = 10.5 Hz, CH₂CH₂CHFCHF); **HRMS** (ESI, +ve) $C_{19}H_{20}O_{3}F_{2}Na^{+}$ [MNa⁺] requires m/z 357.1273, found 357.1263.

(5S,6R)-5-((Benzyloxy)methyl)-6-fluoro-5,6,7,8-tetrahydronaphtho[2,3-d][1,3]dioxole (12)



Obtained as a side-product during the synthesis of difluoroalkane 11. Data for 12: pale yellow oil; $[\alpha]_{D}$ -19.8 (c 0.561, CHCl₃); **IR** (DCM) v_{max} (cm⁻¹): 3067, 3033, 2880, 2181, 2091, 1869, 1737, 1485, 1453, 1360, 1330, 1225, 1113, 1092, 1038; ¹H NMR (400 MHz, CDCl₃) & 7.37–7.28 (m, 5H, ArH), 6.84 (s, 1H, ArH), 6.58 (s, 1H, ArH), 5.89 (ddd, J = 1.3, 3.9, 12.3 Hz, 1H, OCH₂O), 5.20 (dddd, J = 2.5, 3.9, 6.5, 49.2, 1H, CHF), 4.61 (d, *J* = 12.5 Hz, 1H, PhC<u>H</u>HO), 4.58 (d, *J* = 12.5 Hz, 1H, PhCH<u>H</u>O), 3.87 (ddd, *J* = 0.9, 5.7, 9.1 Hz, 1H, CHHOBn), 3.83 (dd, J = 6.9, 9.1 Hz, 1H, CHHOBn), 3.15 (ddd, J = 6.1, 10.5, 27.4 Hz, 2H, CHCH₂OBn), 2.95 (ddd, J = 6.9, 9.7, 16.4 Hz, 1H, CHFCH₂CHH), 2.66 (dt, J = 5.7, 16.4 Hz, 1H, CHFCH₂CHH), 2.24 (ddd, J = 6.1, 13.6, 33.1 Hz, 1H, CHFCHHCH₂), 1.89 (ddddd, J = 2.5, 6.8, 9.3, 1.2513.6, 34.2 Hz, 1H, CHFCHHCH₂); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 146.3 (d, J = 3.1 Hz), 138.4, 129.3, 128.5, 127.8, 127.7, 127.6, 127.5, 108.5, 108.0, 100.9, 89.1 (d, 173.5Hz), 73.5, 71.1 (d, J= 7.3 Hz), 42.8 (d, J = 19.4 Hz), 27.1 (d, J = 20.7 Hz), 25.3 (d, J = 7.5 Hz); ¹⁹F NMR (377 MHz, CDCl₃), δ –195.1 (s, 1F, CH<u>F</u>); ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ –195.1 (m, 1F, CHF); HRMS (ESI, +ve) C₁₉H₁₉O₃FNa⁺ [MNa⁺] requires *m/z* 337.1210, found 337.1206.

(2*R*,3*R*)-5-(Benzo[*d*][1,3]dioxol-5-yl)-2,3-difluoropentan-1-ol (13)

Compound 11 (28.1 mg, 0.084 mmol) was dissolved in methanol (2 mL), and Pd/C (10%, 24.5 mg) was added to this solution. The mixture was stirred under H₂ atmosphere at r.t. for 18 h. The mixture was filtered through Celite and washed with methanol. The filtrate was concentrated under reduced pressure to yield the title compound as a sticky white solid (19.8 mg, 97%); $[\alpha]_{\rm D}$ +26.3 (c 1.00, CHCl₃); IR (neat) $v_{\rm max}$ (cm⁻¹) 3338, 2930, 2060, 1997, 1719, 1608, 1491, 1444, 1360, 1247, 1190, 1098, 1039; ¹H NMR (400 MHz, $CDCl_3$) δ 6.74 (d, J = 7.9 Hz, 1H, ArH), 6.69 (d, J = 1.6 Hz, 1H, ArH), 6.65 (dd, J = 1.6, 7.9 Hz, 1H, ArH), 5.93 (s, 1H, OCH₂O), 4.59 (dddt, J = 3.5, 9.6, 22.8, 47.5 Hz, CHFCH₂OH), 4.52 (ddddd, J = 3.5, 3.5, 6.7, 23.0, 48.0 Hz, CHFCHFCH₂OH), 3.91 (ddd, *J* = 6.7, 12.6, 19.1 Hz, 1H, CHHOH), 3.82

(ddd, J = 3.5, 12.6, 25.6 Hz, 1H, C<u>H</u>HOH), 2.78 (ddd, J = 5.2, 8.6, 13.8 Hz, 1H, C<u>H</u>HCH₂CHF), 2.66 (dt, J = 8.2, 13.8 Hz, 1H, CH<u>H</u>CH₂CHF), 2.13 (ddd, J = 4.8, 9.1, 13.8, 23.0 Hz, 1H, CH₂C<u>H</u>HCHF), 1.82 (m, 1H, CH₂CH<u>H</u>CHF); ¹³C{¹H} **NMR** (101 MHz, CDCl₃) δ 147.9, 146.1, 134.5, 121.4, 109.0, 108.5, 101.0, 93.7 (dd, J = 19.7, 177.4 Hz), 90.6 (dd, J = 19.7, 175.1 Hz), 61.8 (dd, J = 7.2, 23.9 Hz), 32.4 (dd, J = 4.7, 21.2 Hz), 30.8 (d, J = 4.4 Hz); ¹⁹FNMR (377 MHz, CDCl₃), δ -200.1 (d, J = 11.4 Hz, CH₂CH₂CH<u>F</u>), -204.9 (d, J = 11.4 Hz, CH₂CH₂CHFCH<u>F</u>); ¹⁹F{¹H} **NMR** (377 MHz, CDCl₃), δ -200.1 (m, 1F, CH₂CH₂CH<u>F</u>), -204.9 (m, 1F, CH₂CH₂CHFCH<u>F</u>); **HRMS** (ESI, +ve) C₁₂H₁₄F₂O₃Na⁺ [MNa⁺] requires *m/z* 267.0803, found 267.0808.

Ethyl (2R,3S)-5-(benzo[d][1,3]dioxol-5-yl)-2,3-dihydroxypentanoate (16)

OH O A mixture of AD-mix-a (1.4 g), tert-butanol (1.5 mL) and water (1.5 mL) OEt was stirred at r.t for 30 min. The mixture then was cooled to 0 °C, and ester ŌН (0.24 g, 0.97 mmol) was added, followed by the addition 15 methanesulfonamide (95 mg, 0.97 mmol), the reaction mixture was stirred at 0 °C for 48 h. The reaction was quenched with addition sodium sulfite (1.5 g, 12 mmol), and stirred at r.t for 1 h. The mixture was extracted with ethyl acetate $(4 \times 100 \text{ mL})$ and dichloromethane $(4 \times 100 \text{ mL})$, and the combined organic layers were dried (MgSO₄) and concentrated onto silica. The residue was subjected to flash chromatography eluting with 3:7 ethyl acetate/hexane to give the title compound as a white waxy solid (167 mg, 61%); $[\alpha]_{\rm D}$ -24.5 (c 0.388, CHCl₃); m.p. 82.4–82.7 °C; IR (neat) $v_{\rm max}$ (cm⁻¹) 3453, 3350, 2903, 2779, 1844, 1727, 1384, 1502, 1439, 1332, 1292, 1238, 1190, 1110, 1068, 1036; ¹**H** NMR (400 MHz, CDCl₃) δ 6.72 (d, J = 7.9 Hz, 1H, ArH), 6.70 (d, J = 1.5 Hz, 1H, ArH), 6.65 (dd, J = 1.5, 7.9 Hz, 1H, ArH), 5.91 (s, 2H, OCH₂O), 4.27 (ddd, J = 2.1, 7.1, 13.9 Hz, 2H, CH₂CH₃),4.08 (d, J = 1.8 Hz, 1H, CHOHC=O), 3.89 (ddd, J = 1.8, 4.7, 8.3 Hz, 1H, CHOHCHOH), 2.74 (ddd, J = 5.6, 9.2, 14.0 Hz, 1H, PhCHHCH₂), 2.63 (ddd, J = 7.2, 8.8, 14.0 Hz, 1H, PhCHHCH₂), 1.96–1.80 (m, 2H, CH₂CH₂CHOH), 1.30 (t, J = 7.1 Hz, 3H, CH₂CH₃), ¹³C{¹H} NMR (101 MHz, CDCl₃) 173.6, 147.8, 145.8, 135.5, 121.3, 109.0, 108.3, 100.9, 73.3, 71.8, 62.3, 35.7, 31.8, 14.3; HRMS (ESI, +ve) $C_{14}H_{18}O_6Na^+$ [MNa⁺] requires *m/z* 305.0996, found 305.0991.

Ethyl (4*R*,5*S*)-5-(2-(benzo[*d*][1,3]dioxol-5-yl)ethyl)-1,3,2-dioxathiolane-4-carboxylate 2,2dioxide (17)



Step 1: Thionyl chloride (110 μ L, 1.52 mmol) was added to a solution of the diol **16** (212 mg, 0.751 mmol) and pyridine (180 μ L, 2.25 mmol) in dichloromethane (8.75 mL) at 0 °C, and the resulting mixture was stirred at 0 °C for 5 h. Saturated aqueous copper sulfate (30 mL) was added, and the

reaction was extracted with dichloromethane (3×70 mL). The combined organic layers were dried (MgSO₄) and concentrated to give the corresponding cyclic sulfite as a yellow/green oil which was used in the next step without further purification.

Step 2: The crude cycle sulfite (~0.751 mmol) was dissolved in acetonitrile (5.8 mL) and dichloromethane (5.8 mL) and the solution was cooled to 0 °C. Sodium metaperiodate (313 mg, 0.45 mmol), ruthenium chloride hydrate (~10 mg) and water (8.5 mL) were added, and the mixture was stirred at 0 °C for 16 h. Diethyl ether (150 mL) was added, and the mixture was washed with water (2 × 75 mL), saturated aqueous sodium bicarbonate (2 × 75 mL) and brine (2 × 75 mL). The organic layer was dried with MgSO₄ and concentrated to give a mixture of the title compound as a dark orange oil (149 mg, 59%) and unreacted cyclic sulfite intermediate (69.8 mg, 31%). Data for **17**: $[\alpha]_D$ –70.4 (*c* 0.27, CHCl₃); **IR** (CHCl₃) v_{max} (cm⁻¹) 2591, 2544, 2361, 2160, 1767, 1491, 1445, 1395, 1208, 1020; ¹H NMR (400 MHz, CDCl₃) δ 6.75 (d, *J* = 7.8 Hz, 1H, ArH), 6.67 (d, *J* = 1.4 Hz, 1H, ArH), 6.64 (dd, *J* = 1.4, 7.8 Hz, 1H, ArH), 5.94 (s, 2H, OCH₂O), 4.90 (ddd, 1H, *J* = 4.0, 7.2, 8.4 Hz, CHSO₄CH), 4.86 (t, 1H, *J* = 7.2 Hz, CHSO₄CH), 4.32 (dd, *J* = 7.2, 14.2 Hz, 2H, OCH₂CH₃), 2.85 (ddd, H, *J* = 5.4, 8.6, 14.0 Hz, 1H, PhCHHCH₂), 2.71 (dt, H, *J* = 8.3, 14.0 Hz, 1H, PhCHHCH₂), 2.35 2.18 (m, 2H, PhCH₂CH₂), 1.32 (t, *J* = 7.2 Hz, 1H, OCH₂CH₃), ¹³C{¹H}</sup> NMR (101 MHz, CDCl₃) 164.8, 148.1, 146.5, 132.7, 121.5, 108.9, 108.7, 101.2, 83.1, 79.9, 63.5, 35.1, 30.8, 14.1.

Ethyl (2S,3S)-5-(benzo[d][1,3]dioxol-5-yl)-2-fluoro-3-hydroxypentanoate (18)



A mixture of tetra-*N*-butylammonium fluoride (1M solution in THF, 1.5 mL, 1.5 mmol) and acetonitrile (12.5 mL) was dried over 4Å molecular sieves then added via cannula to an ice-cold flask containing a mixture of cyclic

sulfate 17 and sulfite precursor (~2:1, ~0.64 mmol). The resulting mixture was stirred at r.t. for 4.5 h. The mixture was concentrated to give a clear brown oil. This residue was dissolved in a mixture of sulfuric acid (0.12 mL), water (22 μ L) and tetrahydrofuran (22 mL), and the resulting solution was stirred at r. t. for 19 h. Brine (50 mL) and water (50 mL) were added, and the mixture was extracted

with dichloromethane (3 × 150 mL). The combined organic layers were dried (MgSO₄) and concentrated onto silica. The crude product was subjected to flash chromatography eluting with 1:4 ethyl acetate/hexane to give the title compound as a clear yellow liquid (58 mg, 28%); $[\alpha]_D$ –31 (*c* 0.86, CHCl₃); **IR** (CHCl₃) v_{max} (cm⁻¹) 3512, 2931, 2661, 1740, 1611, 1491, 1444, 1374, 1246, 1095, 1039; ¹H NMR (400 MHz, CDCl₃) δ 6.72 (d, *J* = 7.9 Hz, 1H, ArH), 6.68 (s, 1H, ArH), 6.64 (d, *J* = 7.9 Hz, 1H, ArH), 5.91 (s, 2H, OC<u>H</u>₂O), 4.83 (dd, *J* = 4.0, 48.3 Hz, 1H, C<u>H</u>F), 4.27 (ddd, *J* = 2.4, 7.0, 14.1 Hz, 2H, OC<u>H</u>₂CH₃), 3.99 (ddd, *J* = 4.0, 9.6, 17.5 Hz, 1H, CH₂HCOH), 2.79 (ddd, H, *J* = 5.2, 8.8, 14.0 Hz, 1H, PhC<u>H</u>HCH₂), 2.63 (dt, H, *J* = 8.1, 14.0 Hz, 1H, PhCH<u>H</u>CH₂), 2.36 (s, 1H, CH₂HCO<u>H</u>), 1.93–1.76 (m, 2H, PhCH₂C<u>H</u>₂), 1.29 (t, *J* = 7.1 Hz, 1H, OCH₂C<u>H</u>₃), ¹³C{¹H} NMR (101 MHz, CDCl₃) 168.1 (d, *J* = 23.9 Hz), 147.8, 145.9, 135.0, 121.3, 109.0, 108.3, 100.9, 91.3 (d, *J* = 187.7 Hz), 70.7 (d, *J* = 21.8 Hz), 61.9, 33.3 (d, *J* = 4.7 Hz), 31.3, 14.2; ¹⁹F NMR (377 MHz, CDCl₃), δ –198.9 (s, 1F, CH<u>E</u>); ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ –198.9 (dd, *J* = 17.7, 48.3, 1F, CH<u>E</u>); **HRMS** (ESI, +ve) Cl₄H₁₇FO₅Na⁺ [MNa⁺] requires *m/z* 307.0952, found 307.0948.

Ethyl (2*R*,3*R*)-5-(benzo[*d*][1,3]dioxol-5-yl)-2,3-difluoropentanoate (19)

A mixture of fluorohydrin 18 (55 mg, 0.19 mmol) and a solution of Deoxo-FluorTM in toluene (2.7 M, 0.71 mL, 1.9 mmol) was stirred at 70 °C in a Teflon vessel for 20 h. The mixture was cooled to 0 °C, diluted with dichloromethane, and concentrated onto silica. The crude product was subjected to flash chromatography eluting with 1:4 ethyl acetate and hexane to give the title compound as a pale yellow oil (29 mg, 52 %); $[\alpha]_{\rm D}$ +31.3 (c 0.560, CHCl₃); **IR** (DCM) $\nu_{\rm max}$ (cm⁻¹) 2938, 1767, 1491, 1444, 1374, 1299, 1246, 1123, 1038; ¹**H NMR** (400 MHz, CDCl₃) δ 6.75 (d, J = 7.8 Hz, 1H, ArH), 6.69 (s, 1H, ArH), 6.65 (d, *J* = 7.8 Hz, 1H, ArH), 5.93 (s, 1H, OCH₂O), 4.84 (ddddd, *J* = 2.2, 3.7, 9.8, 25.1, 46.2, 1H, CH₂CHFCHF), 4.83 (ddd, J = 2.2, 28.6, 47.3, 1H, CH₂CHFCHF), 4.33 (d, J = 7.1 Hz, 1H, CH₃C<u>H</u>HO), 4.29 (d, J = 7.1 Hz, 1H, CH₃CH<u>H</u>O), 2.79 (ddd, J = 5.1, 8.9, 14.0 Hz, 1H, CH<u>H</u>CH₂CHF), 2.66 (ddd, J = 8.1, 8.1, 14.0 Hz, 1H, C<u>H</u>HCH₂CHF), 2.23 (ddd, J = 5.2, 13.6, 23.9Hz, 1H, CH₂CHHCHF), 1.98–1.81 (m, 1H, CH₂CHHCHF), 1.32 (t, *J* = 7.1, 7.1 Hz, 1H, OCH₂CH₃); $^{13}C{^{1}H} NMR$ (101 MHz, CDCl₃) δ 166.8 (dd, 4.5, 24.3 Hz), 147.9, 146.2, 137.6, 134.1, 128.4, 109.0, 108.5, 101.0, 90.9 (dd, J = 20.2, 179.4 Hz), 89.0 (dd, J = 20.8, 194.7 Hz), 62.3, 32.1 (dd, J = 20.8, 194.7 Hz), 62.3, 194.7 Hz), 62.3, 194.7 Hz), 62.3, 194.7 Hz), 62.3, 194.7 Hz), 194.7 Hz), 194.7 Hz), 194.7 Hz 4.3, 21.4 Hz), 30.8 (d, J = 4.8 Hz), 14.2; ¹⁹F NMR (377 MHz, CDCl₃), $\delta - 197.1$ (s, 1F, CH₂CH₂CH<u>F</u>), -208.2 (s, 1F, CH₂CHFCH<u>F</u> CH₂CHFCH<u>F</u>); ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ -197.1 (d, J = 10.1 Hz, CH₂CH₂CHF), -208.1 (d, J = 10.1 Hz, CH₂CHFCHF); HRMS (ESI, +ve) C₁₄H₁₆O₄F₂Na⁺ [MNa⁺] requires *m/z* 309.0909, found 309.0903.

Ethyl (2*R*,3*R*)-5-(benzo[*d*][1,3]dioxol-5-yl)-2,3-difluoropentanoate (20)



A solution of water (910 μ L), HCl (10.2 M, 29 μ L, 0.296 mmol) and 1,4dioxan (0.25 mL) was added to the difluorinated-ester **19** (14.1 mg, 0.049). The mixture was heated to 100 °C for 6 h. The mixture was diluted with water,

and aqueous sodium bicarbonate solution was added until the pH was basic. The mixture was extracted with ethyl acetate (3 × 5 mL). The aqueous layer was acidified with HCl (2M), then extracted with ethyl acetate (3 × 5 mL). All of the organic layers were combined, dried (MgSO₄) and concentrated under reduced pressure to give the title compound as a pale yellow oil (12.6 mg, 99%); $[\alpha]_{D}$ +27.8 (c 0.630, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.75 (d, J = 7.9 Hz, 1H, ArH), 6.69 (s, 1H, ArH), 6.65 (d, J = 7.8 Hz, 1H, ArH), 5.94 (s, 1H, OCH₂O), 4.89 (ddd, J = 1.4, 29.4, 47.0, 1H, CHFCOOH), 4.87 (ddddd, *J* = 1.4, 3.7, 8.8, 25.2, 45.6, 1H, CHFCHFCOOH), 2.80 (ddd, *J* = 5.2, 8.8, 14.0 Hz, 1H, CHHCH₂CHF), 2.67 (ddd, J = 8.0, 8.0, 14.0 Hz, 1H, CHHCH₂CHF), 2.28 (dddd, J =4.5, 8.8, 13.5, 27.2 Hz, 1H, CH₂C<u>H</u>HCHF), 2.01–1.84 (m, 1H, CH₂CHHCHF); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 170.9 (dd, 4.1, 25.7 Hz), 148.0, 146.2, 134, 121.5, 109.0, 108.5, 101.1, 90.7 (dd, J = 19.9, 178.8 Hz), 88.5 (dd, J = 21.2, 194.0 Hz), 32.0 (dd, J = 4.0, 21.3 Hz), 30.7 (d, J = 4.7 Hz); ¹⁹F NMR (377 MHz, CDCl₃), δ –197.0 (s, 1F, CH₂CH₂CHF), –208.5 (s, 1F, ¹⁹F{¹H} NMR (377 MHz, CDCl₃), δ –197.0 (d, J=9.9 Hz, CH₂CHFCHF CH₂CHFCHF); CH₂CH₂CHF), -208.5 (d, J = 10.1 Hz, CH₂CHFCHF); **HRMS** (ESI, +ve) C₁₂H₁₂O₄F₂Na⁺ [MNa⁺] requires *m/z* 281.0596, found 281.0596.

3. Attempted oxidation of difluorinated alcohol 13



Entry	Conditions	Outcome
1	NaIO ₄ (4.0 equiv.), RuCl ₃ (0.1 equiv.), MeCN, DCM, H ₂ O, r.t., 18 h.	13 (51%)
2	NaO ^t Bu (3.0 equiv.), benzene, O ₂ , r.t., 24 h.	13 (95%)
3	PDC (6.0 equiv.), DMF, r.t., 17 h.	13 (36%) + multiple other products
4	Jones reagent 4 <i>N</i> –8 <i>N</i> (excess), acetone, 0 °C to r.t., 2– 8h.	13 (18%) + multiple other products
5	Solution H ₅ IO ₆ /CrO ₃ (0.44M, 2.9 equiv.) MeCN/H ₂ O (3/1), 3–6h r.t., then 50 °C.	13 (77%) + multiple other products

3. Measurement of enantiopurity

Mosher ester analysis²



Alcohol **10** (0.006 mmol) and either (*S*)- or (*R*)-Mosher acid (0.030 mmol) were dissolved in dry DCM (350 μ L). *N*,*N*'-dicyclohexylcarbodiimide (0.020 mmol) was added, followed by 4-dimethylaminopyridine (0.020 mmol), and the mixture was stirred at r.t. until the reaction was

complete as observed by TLC (typically 24 h). The mixture was filtered and concentrated under reduced pressure. The residue was analysed by ${}^{19}F{}^{1}H$ NMR, focusing on and integrating signals in the CHF region.

Chiral HPLC analysis

The enantiomeric purity of **2** was assessed via analytical HPLC, using a Chiralcel OD-H column and PDA detector (300nm), with a solvent system of propanol/hexane (5/95). Two peaks were observed, with retention times of 16.9 min and 17.6 min (see below).



4. Reproductions of NMR spectra

¹H NMR (600 MHz, CDCl₃) of **2**



¹⁹F NMR (565MHz, CDCl₃) of **2**



¹⁹F {¹H} NMR (565 MHz, CDCl₃) of **2**



--195.87 --195.89



¹³C{¹H} NMR (101 MHz, CDCl₃) of 8











¹³C{¹H} NMR (101 MHz, CDCl₃) of 10









¹³C{¹H} NMR (101 MHz, CDCl₃) of 11





^{19}F NMR (377 MHz, CDCl₃) of 11









¹³C{¹H} NMR (101 MHz, CDCl₃) of 12







S-23



¹³C{¹H} NMR (101 MHz, CDCl₃) of **13**







¹⁹F{¹H} NMR (377 MHz, CDCl₃) of 13





¹H NMR (400 MHz, CDCl₃) of **17** + cyclic sulfite precursor



¹³C{¹H} NMR (101 MHz, CDCl₃) of 17 + cyclic sulfite precursor





¹³C{¹H} NMR (101 MHz, CDCl₃) of 18







¹⁹F {¹H} NMR (377 MHz, CDCl₃) of 18





-198.89



¹³C{¹H} NMR (101 MHz, CDCl₃) of 19











¹³C{¹H} NMR (101 MHz, CDCl₃) of 20





^{19}F {¹H} NMR (377 MHz, CDCl₃) of 20



5. Photostability assessment

Piperine (1) and analog 2 were dissolved in ethanol in separate clear glass vials. Both vials were exposed to sunlight, side-by-side, for 2.5 hours. The solutions were then concentrated, and each residue was analysed by ¹H NMR (see below). Unfortunately the NMR solvent (CDCl₃) seemed to have been contaminated with ethyl acetate in some instances.





6. Conformational analysis of compound 2

Accurate NMR *J*-values for **2** were obtained by simulating the ¹H NMR spectrum of this compound using the Daisy module of the Bruker TopSpin software (see enlargements below).



Density functional theory calculations were carried out in Gaussian 09 using a Linux-based computational cluster (Katana, UNSW). Starting conformers of **2** were generated in IQmol by systematically rotating three dihedral angles (F–C–C–F, F–C–C=O and O=C–N–C) in 120° increments using the *rotbond.pl* script developed by Prof. Michelle Coote. All of the starting conformers were then geometry optimised and their energies calculated at the M06-2X level of theory with the 6-311+G(d,p) basis set, and with SMD chloroform solvent. Duplicate final structures were removed. NMR coupling constants were calculated for the lowest energy final structures using the GIAO method at B3LYP/6-311+G(d,p).

Geometry	3D structure	Relative energy (kJ/mol) in chloroform
2a	A A A A A A A A A A A A A A A A A A A	0.00

2b		+1.7
2c		+5.1
2d		+5.6
2e		+7.6
2f	A CONTRACTOR	+8.0
2g		+8.3
2h		+8.8
2i		+8.9

2j		+9.4
2k	X	+11.8
21		+14.0
2m	XXXXXXX	+15.1
2n	A A A A A A A A A A A A A A A A A A A	+15.6
20	and the second	+15.7
2р		+16.3

2q	+18.1

7. Biological assays

AChE inhibition assay³

5,5'-Dithiobis(2-nitrobenzoic acid) (DTNB), acetylthiocholine (ATCI) and acetylcholinesterase from Electrophorus electricus (AChE) were purchased from Sigma Aldrich. The absorbance at 415 nm was measured using an iMark BioRad plate reader.

Three buffers were used in the assay at pH = 8: buffer A (50 mM Tris-HCl); buffer B (50 mM Tris-HCl and 0.1% bovine serum albumin); and buffer C (50 mM Tris-HCl, 100 mM NaCl and 20 mM MgCl₂). Fresh solutions were prepared for each experiment: ATCl (15 mM in milli-Q water), DTNB (3 mM in buffer C) and AChE (0.26 U/ml in buffer A).

In each 96-well plate, DTNB (100 μ L, 3 mM), AChE (20 μ L of 0.26 U/mL), buffer B, test compounds (20 μ L) at different concentrations or buffer A (20 μ L, non-specific) were added. The plate was then mixed and incubated for 15 min at r.t. The initial absorbances were measured at 415 nm, then ATCl (20 μ L) was added, and absorbance was measured at 415 nm every 2 min for 20 reads. The absorbance versus time was plotted for each well, giving a linear relation whose slope was were used to calculate the AChE inhibition activity using the equation below, where *S*_{NS} is the slope of the non-specific control well (buffer A) and S_X is the slope of a test well.

$$\% AChEI = \frac{(S_{NS} - S_X)}{S_{NS}} \times 100\%$$

BACE-1 inhibition assay⁴

BACE-1 inhibition assay was carried out using the Reaction Biology Corp assay protocol. Briefly, Recombinant Human BACE-1 (Accession # P56817.1) expressed in Mouse myeloma cell line was dissolved in freshly prepared Reaction Buffer (100 mM sodium acetate, pH = 4.0, with 1% DMSO added before use). This enzyme solution was delivered into the reaction well by acoustic technology (Echo550; LabCyte Inc. Sunnyvale, CA) in the nanolitre range, followed by a solution of the test compound in DMSO. The mixture was incubated for 10 min at r.t.. A solution of Fluorogenic Peptide Substrate IV (R&D System Cat# ES004) in freshly prepared Reaction Buffer was then dispensed into the reaction well to initiate the reaction. The enzyme activity was monitored every 5 min as a timecourse measurement of the increase in fluorescence signal from a fluorescently-labeled peptide substrate for 120 min at r.t.. Data was analysed by taking the slope (signal/time) of the linear portion of measurement, and calculating the slope using Excel. Curve fits were performed using the Graphpad Prism software.

8. References

- De Araújo-Júnior, J. X.; de M. Duarte, C.; Maria, M. C.; Parente, J. P.; Fraga, C. A. M.; Barreiro, E. J. Synthesis of Natural Amide Alkaloid Piperdardine and a New Bioactive Analogue. *Synth. Commun.* 2001, *31*, 117–123.
- (2) Dale, J. A.; Mosher, H. S. Nuclear Magnetic Resonance Enantiomer Reagents. Configurational correlations *via* Nuclear Magnetic Resonance Chemical Shifts of Diastereomeric Mandelate, *O*-Methylmandelate, and α-Methoxy-α trifluoromethylphenylacetate (MTPA) Esters. *J. Am. Chem. Soc.* **1973**, *95*, 512–519.
- (3) Mathew, M.; Subramanian, S. In Vitro Screening for Anti-Cholinesterase and Antioxidant Activity of Methanolic Extracts of Ayurvedic Medicinal Plants Used for Cognitive Disorders. *PLoS One* 2014, 9, e86804.
- (4) Stachel, S. J.; Coburn, C. A.; Steele, T. G.; Jones, K. G.; Loutzenhiser, E. F.; Grego, A. R.; Rajapakse, H. A.; Lai, M.-T.; Crouthamel, M.-C.; Xu, M.; Tugusheva, K.; Linegerger, J. E.; Pietrak, B. L.; Espeseth, A. S.; Shi, X.-P.; Chen-Dodson, E.; Holloway, M. K.; Munshi, S.; Simon, A. J.; Kuo, L.; Vacca, J. P. Structure-Based Design of Potent and Selective Cell-Permeable Inhibitors of Human β-Secretase (BACE-1). *J. Med. Chem.* 2004, *47*, 6447–6450.